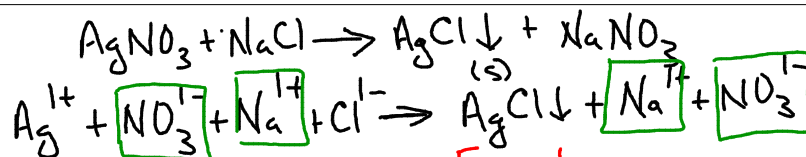


Ch 11
 Sec 11.3
 HW: Sec 11.3 Rev Com 11.3.17
 obj: Write and balance net ionic equations.

Ionic Equations

- An equation w/ all of the ionic compounds written as separate ions.
- * Written for both Single + Double Replacement Rxns.
- * Precipitates are not written in Ionic Form - Double Replacement!!!
- * Products and Reactants that are molecular compounds +/or elements are never written in ionic form.

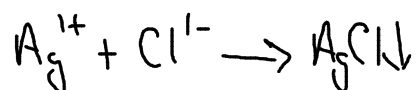
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* Complete Ionic Equation.

* Spectator ions - Not involved in the chemical change.

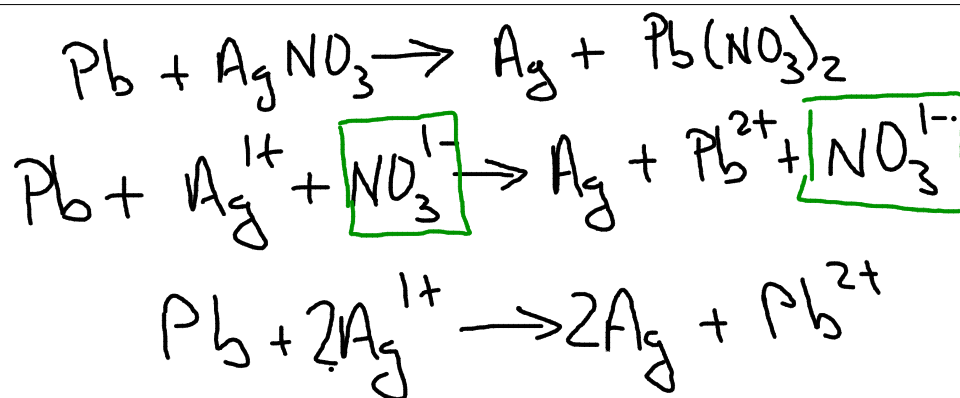
* Net Ionic Equation represents the ions that are involved in the chemical change.



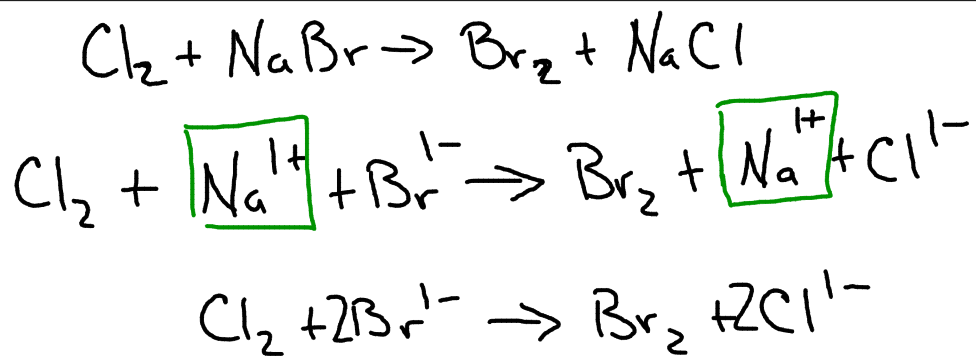
* When balancing net ionic equations balance atoms first. Then balance charges.

* Can only use coefficients.

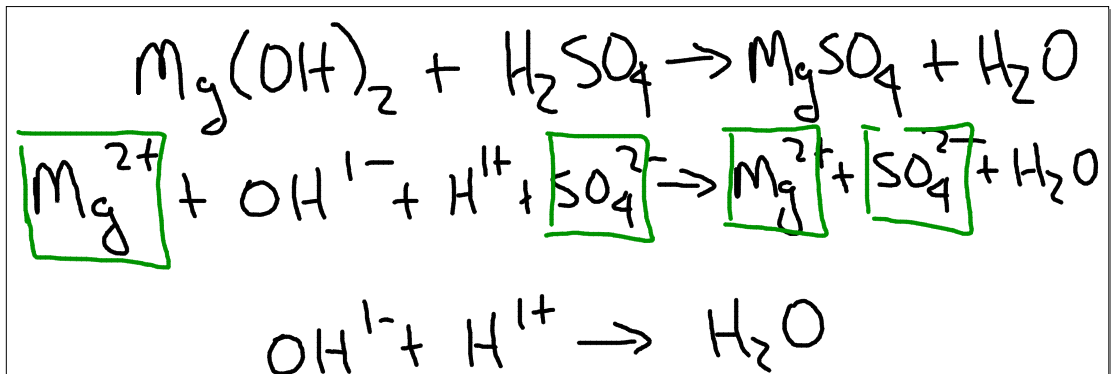
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Dec 2 - 11:41 AM



Dec 2 - 11:48 AM



Nov 28 - 11:05 AM